CURRICULUM MAP – Year 10 COMBINED SCIENCE

Each topic will cover the key enquiry processes, relevant maths skills and cultural capital.

CL = Careers Links which supports the Catholic Social Teaching (CST) strand of Dignity of Work and Participation





KEY Biology Chemistry Physics

HT1:	HT2:	Assessment TD1	HT3:	HT4:	Assessment TD2	HT5	HT6:	Assessment <u>TD3</u>
INTENT	INTENT	Students will	INTENT	INTENT	Students will	INTENT	INTENT	Students
Particle Model of	Atomic Structure	be assessed	Organisation 4.2	Electricity 6.2	be assessed	Infection and	Energy Changes 5.5	will be
Matter 6.3	and the Periodic	by a series of	We learn about the	We learn about	by a series of	Response 4.3	We learn that energy	assessed by
We learn how the	Table 5.1	end of topic	human digestive	electrical charge and	end of topic	We study the	changes are an	a series of
particle model is	We learn how	tests and	system which provides	current in series and	tests and	pathogens which	important part of	end of topic
widely used to	arrangement of	completion of	the body with nutrients	parallel circuits. We	completion of	cause infectious	chemical reactions. The	tests and
predict the	elements in the	the required	and the respiratory	also learn about the	the required	disease in plants and	interaction of particles	completion
behaviour of	modern periodic	practical	system and circulatory	domestic uses of	practical	animals	often involves transfers	of the
solids, liquids and	table can be	booklet.	system that provides it	electricity and how it is	booklet.		of energy due to the	required
gases and how	explained in terms		with oxygen and	supplied.		Link to History	breaking and formation	practical
this has many	of atomic		removes waste. We will			Vaccines, Jenner,	of bonds.	booklet.
applications in	structure which		also learn how the	Link to electrons in 5.1		Smallpox, antibiotics		
everyday life.	provides evidence		plant's transport	CL- Electrical Engineer.		Florey and Chain,	Link to bioenergetics as	
	for the model of a		system is dependent on			antiseptics, Louis	examples of	
Link to KS3 forces,	nuclear atom with		environmental	CST – Creation and		Pasteur, Fleming	endothermic and	
fluids and solids,	electrons in		conditions to ensure	environment – safe			exothermic reactions	
liquids and gases.	energy levels.		that leaf cells are	disposal of batteries.		Link to	CL- Energy Engineer.	
CL- Deep Sea Diver			provided with what	Solidarity – reducing		photosynthesis in 4.4		
Materials	Link to particles		they need for	reliance of fossil fuels.		when looking at		
Researcher,	6.3		photosynthesis.	Options for the poor –		plant diseases	Atomic Structure 6.4	
Engineer, Jeweller.	CL- Research			Use of renewable		CL- Pathologist,	We revisit the structure	
	Scientist, Chemist.		CL- Dietitian	energy resources		Doctor, Lab	of the atom and how	
			Cardiologist, Vet,	where once established		researcher.	this links to ionising	
Cell Biology 4.1	CST – Dignity in		Nutritionist,	take little money to			radiation.	
We explore how	the workplace:		Phlebotomist, Surgeon.	run.		Quantitative		
structural	Extraction of rare					Chemistry 5.3	Recap bonding	
differences	earth resources		CST – Options for the			We use quantitative	Recap DNA from cell	
between types of	for a developing		poor: Poor diet has an	Chemical Changes 5.4		analysis to determine	biology. Links to the	
cells enables them	market.		impact on health e.g.	We learn about the		the formulae of	teaching of Russia in	
to perform specific	Common good:		heart disease.	extraction of important		compounds and the	Humanities due to the	
functions within	Using chemicals to			resources from the		equations for	radiation poisoning of	
the organism.	make new		Bioenergetics 4.4	earth. We study the		reactions.	Litvinyenko.	
	products to help		We explore how plants	way that some		CL- Pharmacist and		

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Taught before ionising radiation. CL – Animal	improve mankind Bonding, Structure	harness the Sun's energy in photosynthesis in order	elements and compounds react with each other and how	Chemical Engineer.	CL- Medical Physicist, Radiographer, Radiation Protection Practitioner.	
Technician,	and Properties of	to make food. We also	easily they can be			
Biologist, Botanist,	Matter 5.2	explore aerobic vs	'pulled apart'.			
General	We use theories of	anaerobic respiration.				
Practitioner (GP),	structure and		Link to ionic bonding in			
Hospital Doctor,	bonding to explain	Link to balancing	5.2			
Pharmacologist.	the physical and	equations from Chem	CL- Quarry Engineer,			
Ŭ	chemical	and links to Geography:	Geoscientist.			
Link to History	properties of	Tropical Rainforests				
HT2 Stem Cells	materials.	CL- Farmer, Gardener,	CST – Dignity in the			
Health and the	CL- Materials	Sports Athlete.	workplace and			
People	Engineer,		participation: Miners in			
	Nanotechnologist,	CST – Solidarity: Our	LEDCs being exploited			
CST –	Research and	need for oxygen and	and poor safety.			
Dignity: Stem cell	Development	how trees and plants				
research form	Manager.	are essential to this but				
embryos		we disturb this balance				
encourages loss of	CST – Peace: Use	through deforestation				
life.	of oil to make	and farming methods.				
Creation and	polymers leads to					
Environment: Life	conflict and					
begins from a	impacts on the					
fertilized egg cell.	environment.					
	Energy C.A.					
	Energy 6.1					
	For the students to look at the					
	different types of					
	energy store,					
	observe in					
	everyday					
	examples how					
	examples now					

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KEY Biology Chemistry Physics

they can be				
transferred and				
calculated and				
investigate the				
main energy				
resources.				
CST – Option for				
the poor –				
Cheaper methods				
of energy				
production.				
Solidarity –				
Sustainable				
resources for the				
future.				
Creation and the				
environment –				
reducing carbon				
footprint				
Link to KS3 energy				
resources. Also				
links to Geography				
CL- Energy				
Engineer,				
Oceanographer,				
Hydrologist.				